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A RECURRENT NOVEL MOTIF FOR 3-DIMENSIONAL MAGNETIC EXCHANGE INTERACTIONS IN CHAIN STRUCTURES CONTAINING ENDO-BIDENTATE α -DI-IMINE LIGANDS: π - π OVERLAP INTERACTIONS

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Abstract

Weakly associated five coordinate monomers based on high-spin trivalent iron α -di-imine complexes are found to exhibit surprisingly strong magnetic exchange interactions that lead to easily discerned 1-and 3-dimensional antiferromagnetic exchange and ordering effects respectively.

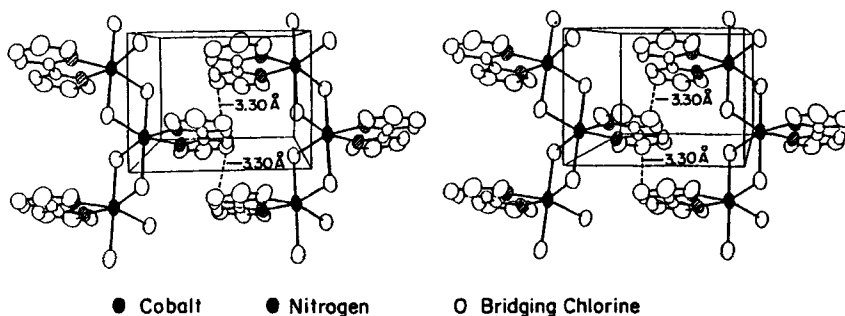
Introduction

For a number of years, we have been interested in the magnetic properties of low dimensionality molecular magnetic materials based on transition metal complexes of simple α -di-imine ligands such 1,10 orthophenanthroline, 2,2'-bipyridine and 2-(pyridin-2-yl) benzothiazole. The synthetic strategy is to prepare formally coordinatively unsaturated materials by using 1:1 metal to ligand preparation reaction stoichiometries to obtain materials corresponding to the empirical formula $[M^III LX_2]$ or $[M^III LX_3]$, X=halide or pseudo-halide. The coordinative unsaturation is presumably overcome via *halogen or pseudo-halogen bridging* among the foregoing units leading to articulation of distinctly non-linear or zig-zag chains in some cases.

We find (1) that precisely the foregoing behavior occurs for the $M(2,2'\text{-bipyridine})Cl_2$ and 1,10-orthophenanthroline series. See Figure 1 for the case of $Co(2,2'\text{-bipyridine})Cl_2$ where such chains interleave and inter-act via π - π overlap stacking interactions.

$[Co(bipy)Cl_2]_{\infty}$ Zig-Zag Chain Polymer

Local *cis* $[CoN_2Cl_4]$ Chromophore



$Co-Co-Co < \approx 130^\circ$

Some 20% Overlap of Adjacent Chains

Inter-Chain Stacking Interaction (3.30 \AA C-C Distance) Like Graphite

FIGURE 1 Zig-zag chain structure of $[Co(2,2'\text{-bipyridine})Cl_2]$ with its two bridging chlorine atoms

On the other hand, for the trivalent metal analogues, e.g., $[Fe(2\text{-pyridin-2-yl benzothiazole})Cl_3]$ (A) and the related $[Fe(1,10\text{-phenanthroline})Cl_3]$ and $[Fe(2,2'\text{-bipyridine})Cl_3]$, there is no halogen bridging. That is, a combination of single crystal x-ray diffraction and Mossbauer spectroscopy shows that these ferric materials consist of novel chains or five coordinate monomeric units connected only by π - π stacking interactions of the overlapping di-imine ligands and perhaps weak Cl-Cl contacts. This situation contrasts strongly with that for $Mn(2,2'\text{-bipyridine})Cl_2$, whose single

chlorine atom bridging leads to six coordinate manganese (III) chromophores in zig-zag chains (2) as pictured in Figure 2.

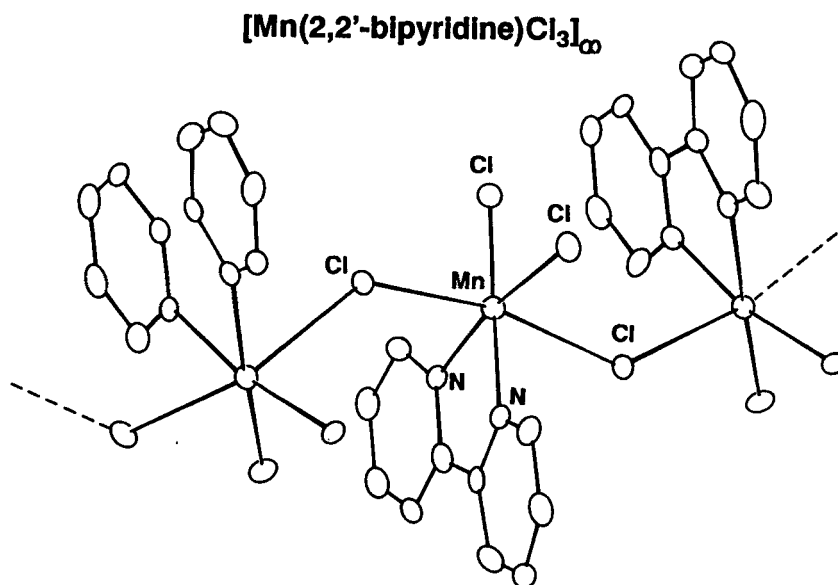


FIGURE 2 Zig-zag chain structure of $[\text{Mn}(2,2'\text{-bipyridine})\text{Cl}_3]$ with its *single bridging chlorine atom* after Perlepes et al (Ref 2)

The individual linear chains and their overlap for compound (A) are shown in Figure 3.

Fe[2-(pyridin-2-yl)benzothiazole]Cl₃
Interleaving Double Chains of 5-Coordinate Monomers

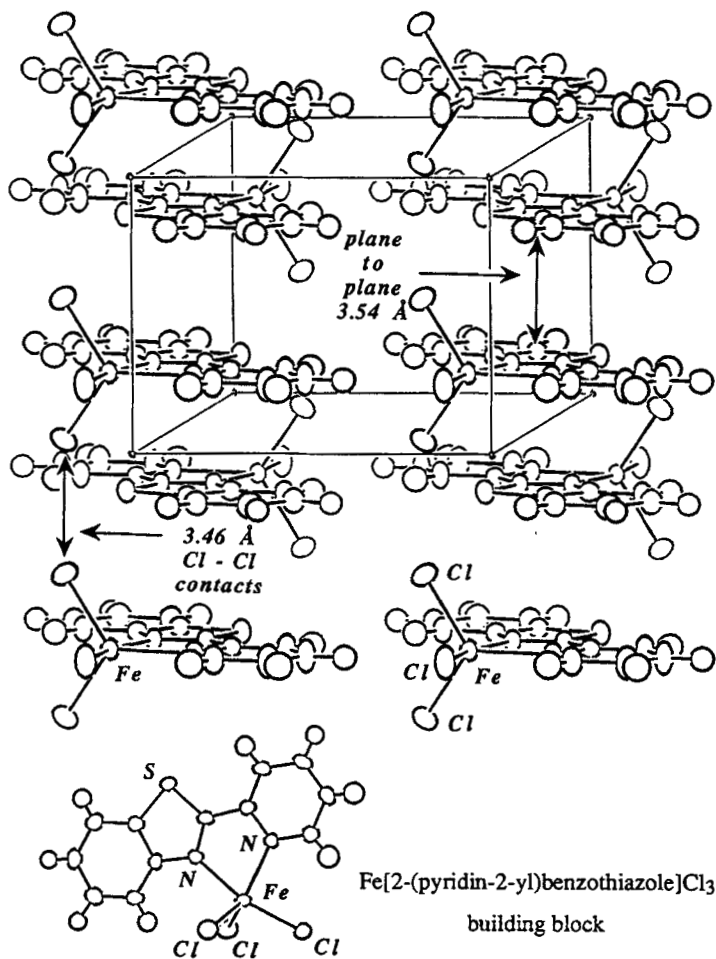


FIGURE 3 Chain structure of [Fe(2-pyridin-2-yl benzothiazole)Cl₃] in which there is no chlorine atom bridging

Nevertheless, these "loosely associated" ferric ion based materials exhibit surprisingly strong magnetic exchange behavior e.g., for (A), $T_N \sim 7.3\text{K}$ and $T(\chi\text{-max-1D}) \sim 13.8\text{K}$) and a rich variety of magnetism. X-ray structure determinations show that the related materials $[\text{Fe}(\text{1,10phenan-throline})\text{Cl}_3\cdot\text{CH}_3\text{OH}]\text{CH}_3\text{OH}$ (B) and $[\text{Fe}(\text{2,2'-bipyridine})\text{Cl}_3\cdot\text{CH}_3\text{OH}]$ (C) exhibit similar details, in particular the $\pi\text{-}\pi$ stacking interactions in addition to hydrogen bonding among Cl and CH_3OH units.

Results

Some a.c. susceptibility data taken at 1 Oe and 125 Hz and bearing on the 3-dimensional magnetic ground state of (A) are given in Figure 4 for a polycrystalline sample.

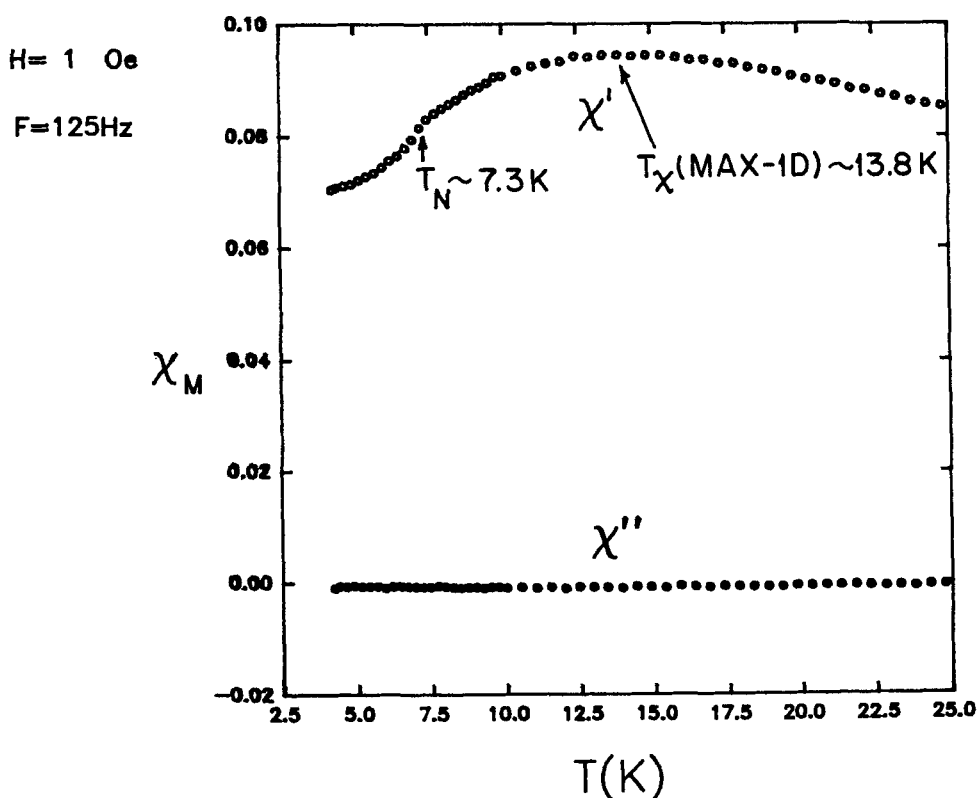


FIGURE 4 A.C. susceptibility data for $[\text{Fe}(\text{2-pyridin-2-yl) benzothiazole})\text{Cl}_3]$

It is clear that there is no out of phase signal (χ''_{m}) thus confirming an uncanted 3D-A.F. ground state for this chain of five coordinate monomers. The temperature dependence of the Mossbauer spectra of (A) suggests $T_{\text{critical}} \sim 7.6\text{K}$ in essential agreement with d.c. and a.c. susceptibility results. The complete details for compounds (A), (B) and (C) including their syntheses, x-ray structures and temperature dependence of Mossbauer spectra will be published elsewhere subsequently (3).

Summary

Suffice it to say for now that rather weak molecular interactions, namely π - π stacking or at most in some cases additional hydrogen bonding can lead to strong magnetic exchange behavior for high-spin trivalent iron mono- α -di-imine complexes. *Supported by the NSF Division of Materials Research.

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